

# **FW-SUT N&A Sunrise Flavor**

Flavor West Manufacturing, LLC.

Version No: **2.2**Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Chemwatch Hazard Alert Code: 2

Issue Date: 04/28/2021
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L.GHS.USA.EN

## SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

## **Product Identifier**

| Product name                  | FW-SUT N&A Sunrise Flavor |
|-------------------------------|---------------------------|
| Synonyms                      | Not Available             |
| Other means of identification | Not Available             |

## Relevant identified uses of the substance or mixture and uses advised against

| Relevant identified | Use according to manufacturer's directions. |
|---------------------|---|
| uses                | ose according to manufacturers directions.  |

# Details of the manufacturer/importer

|                         | •   |  |
|-------------------------|---|--|
| Registered company name | Flavor West Manufacturing, LLC.                       |  |
| Address                 | 29400 Hunco Way, Lake Elsinore CA 92530 United States |  |
| Telephone               | (951) 893-5120  |  |
| Fax                     | (714) 276-1621  |  |
| Website                 | www.FlavorWest.com                                    |  |
| Email                   | Flavor@FlavorWest.com                                 |  |

# **Emergency telephone number**

| Association /<br>Organisation     | Chemwatch |
|-----------------------------------|-----------|
| Emergency telephone numbers       | see below |
| Other emergency telephone numbers | see below |

## **CHEMWATCH EMERGENCY RESPONSE**

| Primary Number | Alternative Number 1 | Alternative Number 2 |
|----------------|----------------------|----------------------|
| 877 715 9305   | +612 9186 1132       | Not Available        |

Once connected and if the message is not in your prefered language then please dial 01

Una vez conectado y si el mensaje no está en su idioma preferido, por favor marque 02

# **SECTION 2 HAZARDS IDENTIFICATION**

## Classification of the substance or mixture

Version No: 2.2 Page 2 of 9 Issue Date: 04/28/2021
Print Date: 04/28/2021

# **FW-SUT N&A Sunrise Flavor**



**GHS Classification** 

Eye Irritation Category 2B

### Label elements

**GHS** label elements

Not Applicable

SIGNAL WORD

WARNING

## Hazard statement(s)

H320

Causes eye irritation

### Precautionary statement(s) Prevention

P264

Wash all exposed external body areas thoroughly after handling.

# Precautionary statement(s) Response

P305+P351+P338

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P337+P313

If eye irritation persists: Get medical advice/attention.

## Precautionary statement(s) Storage

## Precautionary statement(s) Disposal

# SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

## **Substances**

See section below for composition of Mixtures

## **Mixtures**

| CAS No    | %[weight] | Name             |  |
|-----------|-----------|------------------|--|
| 57-55-6   | 90-99     | propylene glycol |  |
| 8001-67-0 | 1-5       | coffee oil       |  |

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

### **SECTION 4 FIRST AID MEASURES**

# Description of first aid measures

| Description of first al | d measures  |
|-------------------------|---|
| Eye Contact             | If this product comes in contact with the eyes:  • Wash out immediately with fresh running water.  • Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.  • Seek medical attention without delay; if pain persists or recurs seek medical attention.  • Removal of contact lenses after an eye injury should only be undertaken by skilled personnel. |
| Skin Contact            | If skin contact occurs:  Immediately remove all contaminated clothing, including footwear.  Flush skin and hair with running water (and soap if available).  Seek medical attention in event of irritation.   |
| Inhalation              | <ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket</li> </ul>                         |

Version No: 2.2 Page 3 of 9 Issue Date: 04/28/2021
Print Date: 04/28/2021

#### **FW-SUT N&A Sunrise Flavor**

|           | <ul> <li>mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor.</li> </ul>   |
|-----------|---|
| Ingestion | <ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious.</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> </ul> |

# Indication of any immediate medical attention and special treatment needed

- Polyethylene glycols are generally poorly absorbed orally and are mostly unchanged by the kidney.
- Dermal absorption can occur across damaged skin (e.g. through burns) leading to increased osmolality, anion gap metabolic acidosis, elevated calcium, low ionised calcium, CNS depression and renal failure.
- Treatment consists of supportive care.

[Ellenhorn and Barceloux: Medical Toxicology]

Propylene glycol is primarily a CNS depressant in large doses and may cause hypoglycaemia, lactic acidosis and seizures.

- The usual measures are supportive care and decontamination (Ipecac/ lavage/ activated charcoal/ cathartics), within 2 hours of exposure should suffice.
- Check the anion gap, arterial pH, renal function and glucose levels.

Ellenhorn and Barceloux: Medical Toxicology

### **SECTION 5 FIREFIGHTING MEASURES**

## **Extinguishing media**

- ▶ Alcohol stable foam.
- ▶ Dry chemical powder.
- ▶ BCF (where regulations permit).
- Carbon dioxide.

## Special hazards arising from the substrate or mixture

Fire Incompatibility

Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may
result

# Advice for firefighters

## Fire Fighting

- Alert Fire Brigade and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- Prevent, by any means available, spillage from entering drains or water course.
- Use water delivered as a fine spray to control fire and cool adjacent area.

# Fire/Explosion Hazard

- ▶ Combustible.
- Slight fire hazard when exposed to heat or flame.
- ▶ Heating may cause expansion or decomposition leading to violent rupture of containers.
- ▶ On combustion, may emit toxic fumes of carbon monoxide (CO).

### **SECTION 6 ACCIDENTAL RELEASE MEASURES**

# Personal precautions, protective equipment and emergency procedures

# Minor Spills

- ▶ Remove all ignition sources.
- Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eyes.
   Control personal contact with the substance, by using protective equipment.

### Chemical Class: alcohols and glycols

For release onto land: recommended sorbents listed in order of priority.

# Major Spills

| SORBENT<br>TYPE | RANK | APPLICATION | COLLECTION | LIMITATIONS |
|-----------------|------|-------------|------------|-------------|
|-----------------|------|-------------|------------|-------------|

## LAND SPILL - SMALL

| cross-linked polymer - particulate | 1 | shovel | shovel    | R, W, SS   |
|------------------------------------|---|--------|-----------|------------|
| cross-linked polymer - pillow      | 1 | throw  | pitchfork | R, DGC, RT |

 Version No: 2.2
 Page 4 of 9
 Issue Date: 04/28/2021

 Print Date: 04/28/2021
 Print Date: 04/28/2021

#### **FW-SUT N&A Sunrise Flavor**

| sorbent clay - particulate  | 2 | shovel | shovel    | R,I, P        |
|-----------------------------|---|--------|-----------|---------------|
| wood fiber - pillow         | 3 | throw  | pitchfork | R, P, DGC, RT |
| treated wood fiber - pillow | 3 | throw  | pitchfork | DGC, RT       |
| foamed glass - pillow       | 4 | throw  | pichfork  | R, P, DGC, RT |

### LAND SPILL - MEDIUM

| cross-linked polymer - particulate | 1 | blower | skiploader | R,W, SS         |
|------------------------------------|---|--------|------------|-----------------|
| polypropylene - particulate        | 2 | blower | skiploader | W, SS, DGC      |
| sorbent clay - particulate         | 2 | blower | skiploader | R, I, W, P, DGC |
| polypropylene - mat                | 3 | throw  | skiploader | DGC, RT         |
| expanded mineral - particulate     | 3 | blower | skiploader | R, I, W, P, DGC |
| polyurethane - mat                 | 4 | throw  | skiploader | DGC, RT         |

#### Legend

DGC: Not effective where ground cover is dense

R; Not reusable

I: Not incinerable

P: Effectiveness reduced when rainy

RT:Not effective where terrain is rugged

SS: Not for use within environmentally sensitive sites

W: Effectiveness reduced when windy

Reference: Sorbents for Liquid Hazardous Substance Cleanup and Control;

R.W Melvold et al: Pollution Technology Review No. 150: Noyes Data Corporation 1988

Moderate hazard.

- Clear area of personnel and move upwind.
- · Alert Fire Brigade and tell them location and nature of hazard.

Personal Protective Equipment advice is contained in Section 8 of the MSDS.

### **SECTION 7 HANDLING AND STORAGE**

# Precautions for safe handling

Safe handling

### → DO N

- ▶ DO NOT allow clothing wet with material to stay in contact with skin
- Avoid all personal contact, including inhalation.
- ▶ Wear protective clothing when risk of exposure occurs.
- Use in a well-ventilated area.
- Prevent concentration in hollows and sumps.
- Other information
- ▶ Material is hygroscopic, i.e. absorbs moisture from the air. Keep containers well sealed in storage.
- ▶ Store in original containers.
- ▶ Keep containers securely sealed.

## Conditions for safe storage, including any incompatibilities

# Suitable container

- Metal can or drum
- Packaging as recommended by manufacturer.
- ▶ Check all containers are clearly labelled and free from leaks.

# Storage

• Glycols and their ethers undergo violent decomposition in contact with 70% perchloric acid. This seems likely to involve formation of the glycol perchlorate esters (after scission of ethers) which are explosive, those of ethylene glycol and 3-chloro-1,2-propanediol being more powerful than glyceryl nitrate, and the former so sensitive that it explodes on addition of water.

# Alcohols

# incompatibility

- are incompatible with strong acids, acid chlorides, acid anhydrides, oxidising and reducing agents.
- reacts, possibly violently, with alkaline metals and alkaline earth metals to produce hydrogen
- react with strong acids, strong caustics, aliphatic amines, isocyanates, acetaldehyde, benzoyl peroxide, chromic acid, chromium oxide, dialkylzincs, dichlorine oxide, ethylene oxide, hypochlorous acid, isopropyl chlorocarbonate, lithium tetrahydroaluminate, nitrogen dioxide, pentafluoroguanidine, phosphorus halides, phosphorus pentasulfide, tangerine oil, triethylaluminium, triisobutylaluminium
- ▶ should not be heated above 49 deg.

## PACKAGE MATERIAL INCOMPATIBILITIES

Not Available

### SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION

#### **FW-SUT N&A Sunrise Flavor**

## **Control parameters**

### OCCUPATIONAL EXPOSURE LIMITS (OEL)

## INGREDIENT DATA

Not Available

### **EMERGENCY LIMITS**

| Ingredient       | Material name                       | TEEL-1   | TEEL-2     | TEEL-3     |
|------------------|-------------------------------------|----------|------------|------------|
| propylene glycol | Propylene glycol; (1,2-Propanediol) | 30 mg/m3 | 1300 mg/m3 | 7900 mg/m3 |

| Ingredient       | Original IDLH | Revised IDLH  |
|------------------|---------------|---------------|
| propylene glycol | Not Available | Not Available |
| coffee oil       | Not Available | Not Available |

### MATERIAL DATA

for propylene glycol:

Saturated vapour concentration @ 20 deg C.= 65.8 ppm, 204.6 mg/m3; i.e higher concentrations can only occur as aerosols or at higher temperatures. Odour Threshold: Practically odourless.

A small number of individuals show skin irritation or sensitisation from repeated or prolonged exposure to propylene glycol. A workplace environmental exposure limit (WEEL) has been established by AIHA and is thought to be protective against systemic effects.

## **Exposure controls**

| •                                   |   |
|-------------------------------------|---|
| Appropriate<br>engineering controls | Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.  The basic types of engineering controls are:  Process controls which involve changing the way a job activity or process is done to reduce the risk.  Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment. |
| Personal protection                 |   |
| Eye and face protection             | <ul> <li>Safety glasses with side shields.</li> <li>Chemical goggles.</li> <li>Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task.</li> </ul>   |
| Skin protection                     | See Hand protection below   |
| Hands/feet protection               | <ul> <li>Wear chemical protective gloves, e.g. PVC.</li> <li>Wear safety footwear or safety gumboots, e.g. Rubber</li> <li>NOTE:</li> <li>The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.</li> <li>Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.</li> </ul>  |
| Body protection                     | See Other protection below  |
| Other protection                    | <ul><li>▶ Overalls.</li><li>▶ P.V.C. apron.</li><li>▶ Barrier cream.</li></ul>  |
| Thermal hazards                     | Not Available   |
|                                     |   |

# Recommended material(s)

## **GLOVE SELECTION INDEX**

Glove selection is based on a modified presentation of the:

## "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the computer-generated selection:

FW-SUT N&A Sunrise Flavor

| Materia | al    | СРІ |
|---------|-------|-----|
| PE/EVA  | AL/PE | A   |

\* CPI - Chemwatch Performance Index

A: Best Selection

# Respiratory protection

Type A-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

Where the concentration of gas/particulates in the breathing zone, approaches or exceeds the "Exposure Standard" (or ES), respiratory protection is required.

Degree of protection varies with both face-piece and Class of filter; the nature of protection varies with Type of filter.

| Required                 | Half-Face  | Full-Face  | Powered Air |
|--------------------------|------------|------------|-------------|
| Minimum                  | Respirator | Respirator | Respirator  |
| <b>Protection Factor</b> | Respirator | Respirator | Respirator  |

Version No: 2.2 Page 6 of 9 Issue Date: 04/28/2021 Print Date: 04/28/2021

### **FW-SUT N&A Sunrise Flavor**

B: Satisfactory; may degrade after 4 hours continuous immersion C: Poor to Dangerous Choice for other than short term immersion NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may

dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

| up to 10 x ES  | A-AUS P2 | -                     | A-PAPR-AUS /<br>Class 1 P2 |
|----------------|----------|-----------------------|----------------------------|
| up to 50 x ES  | -        | A-AUS / Class<br>1 P2 | -                          |
| up to 100 x ES | -        | A-2 P2                | A-PAPR-2 P2 ^              |

<sup>^ -</sup> Full-face

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

## **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

# Information on basic physical and chemical properties

| Appearance                                   | Cloudy pale yellow |   |               |
|--|--------------------|---|---------------|
| Physical state                               | Liquid             | Relative density<br>(Water = 1)         | 1.05          |
| Odour  | Sharp              | Partition coefficient n-octanol / water | Not Available |
| Odour threshold                              | Not Available      | Auto-ignition temperature (°C)          | Not Available |
| pH (as supplied)                             | Not Available      | Decomposition temperature               | Not Available |
| Melting point / freezing point (°C)          | Not Available      | Viscosity (cSt)                         | Not Available |
| Initial boiling point and boiling range (°C) | Not Available      | Molecular weight (g/mol)                | Not Available |
| Flash point (°C)                             | >93.00             | Taste                                   | Gritty        |
| Evaporation rate                             | Not Available      | Explosive properties                    | Not Available |
| Flammability                                 | Not Applicable     | Oxidising properties                    | Not Available |
| Upper Explosive Limit (%)                    | Not Available      | Surface Tension<br>(dyn/cm or mN/m)     | Not Available |
| Lower Explosive Limit (%)                    | Not Available      | Volatile Component<br>(%vol)            | Not Available |
| Vapour pressure (kPa)                        | Not Available      | Gas group                               | Not Available |
| Solubility in water<br>(g/L)                 | Miscible           | pH as a solution (1%)                   | Not Available |
| Vapour density (Air = 1)                     | Not Available      | VOC g/L                                 | Not Available |

## **SECTION 10 STABILITY AND REACTIVITY**

| Reactivity                             | See section 7  |
|--|--|
| Chemical stability                     | <ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> </ul> |
| Possibility of<br>hazardous reactions  | See section 7  |
| Conditions to avoid                    | See section 7  |
| Incompatible materials                 | See section 7  |
| Hazardous<br>decomposition<br>products | See section 5  |

### **SECTION 11 TOXICOLOGICAL INFORMATION**

Version No: 2.2 Page 7 of 9 Issue Date: 04/28/2021 Print Date: 04/28/2021

## **FW-SUT N&A Sunrise Flavor**

| Inhaled                      | The material is not thought to produce respiratory irritation (as classified by EC Directives using animal models).  Nevertheless inhalation of vapours, fumes or aerosols, especially for prolonged periods, may produce respiratory discomfort and occasionally, distress.  Inhalation of vapours may cause drowsiness and dizziness. This may be accompanied by narcosis, reduced alertness, loss of reflexes, lack of coordination and vertigo.  |  |   |  |
|------------------------------|--|--|---|--|
| Ingestion                    | Accidental ingestion of the material may be damaging to the health of the individual.  Tryptamines frequently produce feelings of euphoria, disinhibition, auditory as well as visual disturbances/ distortions and hallucinations, euphoria, dilated pupils, empathy, "feelings of love," and emotional distress. They are often linked to adverse effects including restlessness, agitation, gastrointestinal distress (nausea, vomiting, and diarrhea) and muscle tension.  The most well-known tryptamines are serotonin, an important neurotransmitter, and melatonin, a hormone involved in regulating the sleep-wake cycle.   |  |   |  |
| Skin Contact                 | Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions.  The material may produce moderate skin irritation; limited evidence or practical experience suggests, that the material either:  • produces moderate inflammation of the skin in a substantial number of individuals following direct contact and/or  • produces significant, but moderate, inflammation when applied to the healthy intact skin of animals (for up to four hours), such inflammation being present twenty-four hours or more after the end of the exposure period.  Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The dermatitis is often characterised by skin redness (erythema) and swelling (oedema) which may progress to blistering (vesiculation), scaling and thickening of the epidermis. |  |   |  |
| Еуе                          | Limited evidence exists, or practical experience suggests, that the material may cause eye irritation in a substantial number of individuals and/or is expected to produce significant ocular lesions which are present twenty-four hours or more after instillation into the eye(s) of experimental animals. Repeated or prolonged eye contact may cause inflammation characterised by temporary redness (similar to windburn) of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur.  Irritation of the eyes may produce a heavy secretion of tears (lachrymation).   |  |   |  |
| Chronic                      | There exists limited evidence that shows that skin contact with the material is capable either of inducing a sensitisation reaction in a significant number of individuals, and/or of producing positive response in experimental animals.   |  |   |  |
| FW-SUT N&A                   | TOXICITY IRRITATION  |  |   |  |
| Sunrise Flavor               | Not Available  | Not Available  |   |  |
| propylene glycol             | TOXICITY  Dermal (rabbit) LD50: >2000 mg/kg <sup>[1]</sup> Eye (rabbit): 100 mg - mild  Oral (rat) LD50: 20000 mg/kgd <sup>[2]</sup> Eye (rabbit): 500 mg/24h - mild  Skin(human):104 mg/3d Intermit Mod  Skin(human):500 mg/7days mild  |  | 24h - mild<br>3d Intermit Mod   |  |
| coffee oil                   | TOXICITY  Oral (rat) LD50: 19200 mg/kg* <sup>[2]</sup>   |  | IRRITATION  Not Available   |  |
| Legend:                      | Value obtained from Europe ECHA Registered Substances -     Unless otherwise specified data extracted from RTECS - Regis   |  |   |  |
| FW-SUT N&A<br>Sunrise Flavor | The acute oral toxicity of propylene glycol is very low, and la damage in humans. Serious toxicity generally occurs only at high intake over a relatively short period of time. It would be or supplements, which contain at most 1 g/kg of PG. Cases inappropriate intravenous administration or accidental ingestional toxicity is also low.   | plasma concentrations<br>nearly impossible to re<br>of propylene glycol pois | over 1 g/L, which requires extremely each toxic levels by consuming foods oning are usually related to either |  |
| PROPYLENE GLYCOL             | The material may cause skin irritation after prolonged or repeated exposure and may produce a contact dermatitis (nonallergic). This form of dermatitis is often characterised by skin redness (erythema) and swelling the epidermis. Histologically there may be intercellular oedema of the spongy layer (spongiosis) and intracellular oedema of the epidermis.  The acute oral toxicity of propylene glycol is very low, and large quantities are required to cause perceptible health damage in humans.   |  |   |  |
| COFFEE OIL                   | Group A aliphatic monoesters (fatty acid esters)  According to a classification scheme described by the American Chemistry Council' Aliphatic Esters Panel, Group A substances are simple monoesters derived from a monofunctional alcohol, such as 2-ethylhexyl alcohol (C8-alcohol) or tridecyl alcohol (C13 alcohol) and fatty acids such as palmitic, stearic, oleic or linoleic acid. Metabolism of the parent  |  |   |  |

 Version No: 2.2
 Page 8 of 9
 Issue Date: 04/28/2021

 Print Date: 04/28/2021
 Print Date: 04/28/2021

### **FW-SUT N&A Sunrise Flavor**

esters is expected to yield the corresponding fatty acids and alcohols. The fatty acids are naturally occurring and have a low order of toxicity.

Group A substances are rather lipophilic (log Kow 10-15) in character due to the large number of carbon numbers in the ester molecule (e.g., 24,26, 31 carbons) and have relatively high boiling points.

\* Robertet SDS

| Acute Toxicity                    | 0        | Carcinogenicity             | 0 |
|-----------------------------------|----------|-----------------------------|---|
| Skin<br>Irritation/Corrosion      | 0        | Reproductivity              | 0 |
| Serious Eye<br>Damage/Irritation  | <b>~</b> | STOT - Single<br>Exposure   | 0 |
| Respiratory or Skin sensitisation | 0        | STOT - Repeated<br>Exposure | 0 |
| Mutagenicity                      | 0        | Aspiration Hazard           | 0 |

Legend:

✓ – Data required to make classification available

🗶 – Data available but does not fill the criteria for classification

Data Not Available to make classification

### **CMR STATUS**

Not Applicable

### **SECTION 12 ECOLOGICAL INFORMATION**

## **Toxicity**

### NOT AVAILABLE

| Ingredient       | Endpoint      | Test Duration | Effect        | Value         | Species       | BCF           |
|------------------|---------------|---------------|---------------|---------------|---------------|---------------|
| propylene glycol | Not Available |
| coffee oil       | Not Available |

Propylene glycol is known to exert high levels of biochemical oxygen demand (BOD) during degradation in surface waters. This process can adversely affect aquatic life by consuming oxygen needed by aquatic organisms for survival. Large quantities of dissolved oxygen (DO) in the water column are consumed when microbial populations decompose propylene glycol.

Sufficient dissolved oxygen levels in surface waters are critical for the survival of fish, macro-invertebrates, and other aquatic organisms.

## Persistence and degradability

| Ingredient       | Persistence: Water/Soil | Persistence: Air |
|------------------|-------------------------|------------------|
| propylene glycol | LOW                     | LOW              |

# **Bioaccumulative potential**

| Ingredient       | Bioaccumulation |
|------------------|-----------------|
| propylene glycol | LOW (BCF = 1)   |

## Mobility in soil

| Ingredient       | Mobility       |
|------------------|----------------|
| propylene glycol | HIGH (KOC = 1) |

### **SECTION 13 DISPOSAL CONSIDERATIONS**

### Waste treatment methods

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

# Product / Packaging disposal

- ▶ Reduction▶ Reuse
- r IVense
- RecyclingDisposal (if all else fails)

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use.

 Version No: 2.2
 Page 9 of 9
 Issue Date: 04/28/2021

 Print Date: 04/28/2021
 Print Date: 04/28/2021

### **FW-SUT N&A Sunrise Flavor**

**SECTION 14 TRANSPORT INFORMATION** 

## **Labels Required**

Marine Pollutant

NO

Land transport (DOT): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code / GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

## **SECTION 15 REGULATORY INFORMATION**

### Safety, health and environmental regulations / legislation specific for the substance or mixture

| propylene                |
|--------------------------|
| glycol(57-55-6) is       |
| found on the             |
| following regulatory     |
| lists                    |
| coffee oil(8001-67-0) is |

"US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)", "US - Washington Toxic air pollutants and their ASIL, SQER and de minimis emission values", "US AIHA Workplace Environmental Exposure Levels (WEELs)", "US Spacecraft Maximum Allowable Concentrations (SMACs) for Airborne Contaminants", "US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"

offee oil(8001-67-0) is found on the following regulatory lists

"US - California Permissible Exposure Limits for Chemical Contaminants", "US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory"

| National Inventory               | Status  |
|----------------------------------|---|
| Australia - AICS                 | Y   |
| Canada - DSL                     | Υ   |
| China - IECSC                    | Υ   |
| Europe - EINEC /<br>ELINCS / NLP | Y   |
| Japan - ENCS                     | N (coffee oil)  |
| Korea - KECI                     | Υ   |
| New Zealand - NZIoC              | Υ   |
| Philippines - PICCS              | Υ   |
| USA - TSCA                       | Υ   |
| Legend:                          | Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets) |

## **SECTION 16 OTHER INFORMATION**

### Other information

## Ingredients with multiple cas numbers

| Name       | CAS No                |
|------------|-----------------------|
| coffee oil | 8001-67-0, 84650-00-0 |

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net

The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

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